

The Role of Mechanical Aging and Photoweathering in the Fragmentation of Glitter Microplastics



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Introduction

Microplastic pollution in freshwater poses ecological and health risks, but key processes like fragmentation remain poorly understood. This study investigates how weathering processes such as particle-particle collisions and ultraviolet (UV) aging contribute to the formation of secondary micro- and nanoplastics. Glitter was used as a model microplastic due to it's uniformity and real world relevance as a primary microplastic.

The main goals of this study are to:

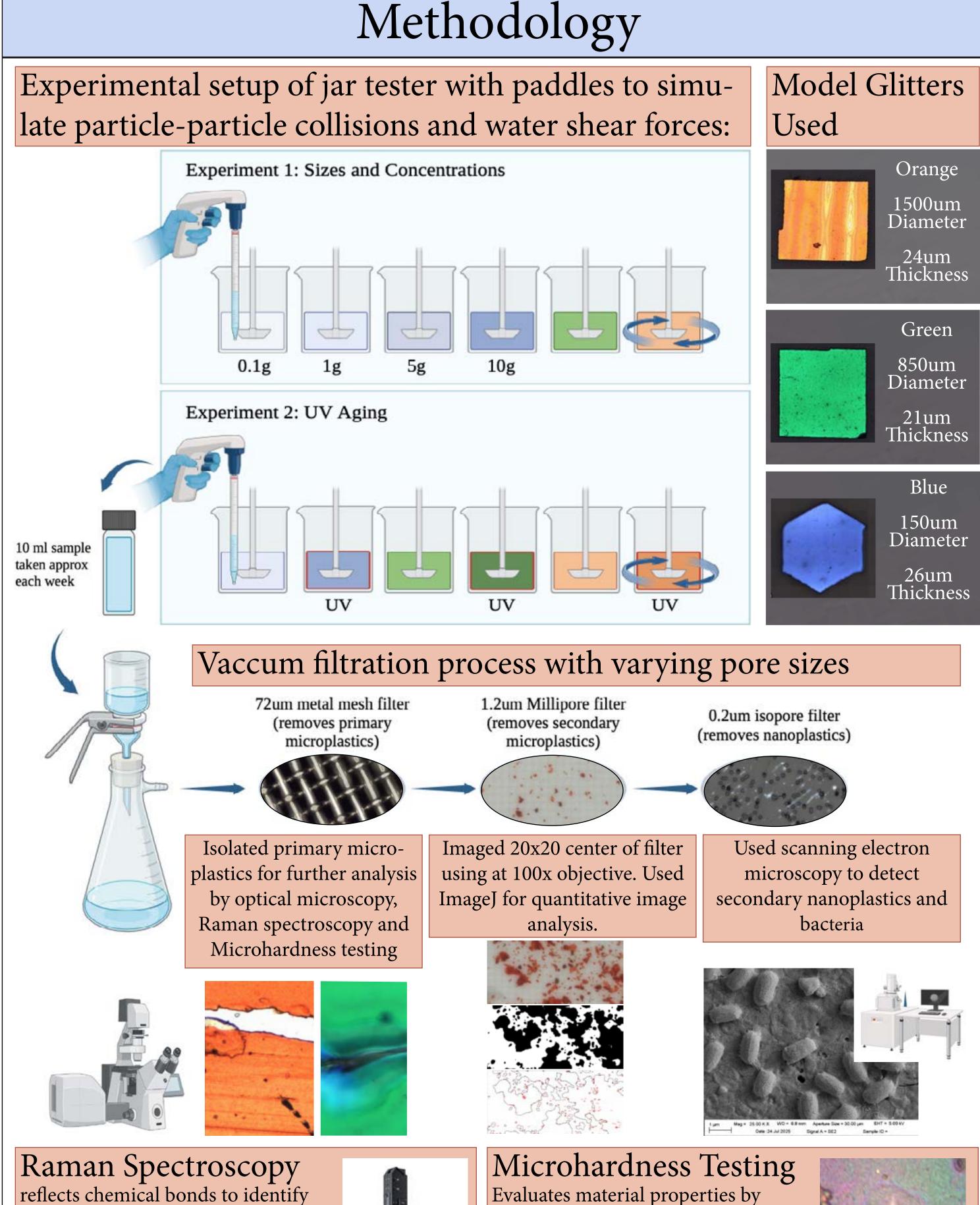
material composition an chemical

changes. Raman spectra were taken

from pristine glitters and after UV

weathering.

- 1) characterize the physical and chemical properties of different size classes and colors of commercial microplastic glitters; and
- 2) test the weathering effects on microplastic glitters by simulating their degradability in wastewater treatment plants and the environment



indentation such as hardness (HT115)

and elastic modulus (Eit). Testing was

done on pristine glitters and after

UV and mechanical weathering.

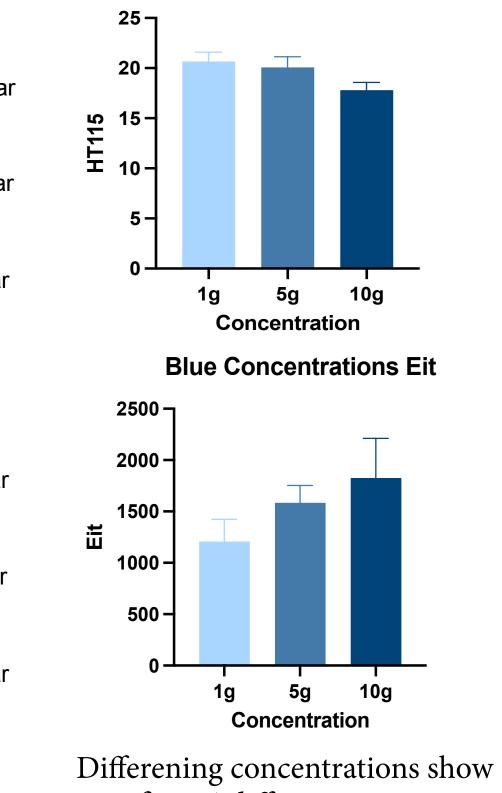
Results Quantitative analysis of microplastics distributed on 1.2um Millipore filter **Different Sizes + Colors Particle Size Distribution**

The histograms show a trend of exponential decay in the size distribution of the microplastics, with most particles in the 0-100 μm range and few above 400 μm. Particle concentration of the blue glitter had no significant effect on size distribution, while glitter size did: the largest glitter (orange, 1500 μm) produced ~10 times more small particles (0–100 μm) than the smallest glitter (blue, $150 \mu m$).

Microhardness Testing HT115 Values Water Shear Water Shear

Orange: Significant* changes in both Eit and HIT115 after UV and mechanical weathering.

Green: Significant* change in HIT115 only; Eit unchanged. Blue: Significant* change in Eit only; HIT115 unchanged.

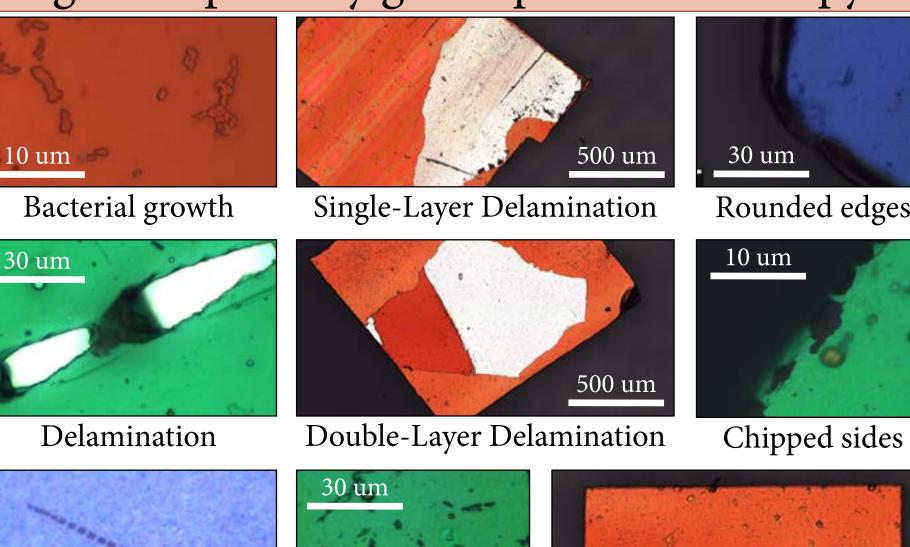


significant* differences in Eit (positive correlation) and HT115 (negative correlation)

*significant = p-value < 0.0001

- CRX 8007, PCLYESTER FILM (Corrected)
- Crange Spectrum Test apic (Corrected)

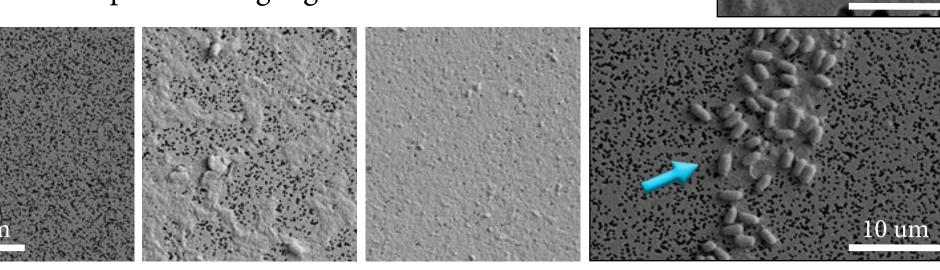
Degraded primary glitter photomicroscopy



Chipped sides Scratches/dents

SEM Photomicroscopy SEM revealed bacterial growth in the wastewater simulation and nanoplastics generated from particle collisions. Bacteria were uniformly rod-shaped ($\sim 1 \ \mu m \times \sim 2 \ \mu m$). Nanoplastics were not uniformly shaped, but generally had sharp edges and triangular shapes ($\sim 0.1 \ \mu m \times \sim 2 \ \mu m$). A biofilm-like layer, likely dead cells and extracellular matrix,

covered much of the filter, hindering nanoplastic detection. More cell material was observed in filters from higher-concentration samples and larger glitter sizes.



2000 1800 1800 1400 1200 1000 800 600 40 Polyester Film *3000 (78.34%) Polyester Film (52.86%) Copper phthalocyanine (72.14%) Polyester Film *3000 (67.42%)

Irgalith Blue (73.46%) Spectrum-011-Spec Data 1 (B+R) (Sub 80).spc (Correct. Copper phthalocyanine (72.14%)

Raman Spectroscopy

Conclusions

Developed a reproducible methodology for filtering and imaging microplastics across size classes.

Track marks

Secondary microplastics (1–2000 µm) formed via mechanical aging collisions showed an exponential decay in size distribution.

Photomicroscopy identified key damage types from glitter-on-glitter collisions.

SEM and optical microscopy confirmed biofilm growth on glitter; higher glitter concentrations and sizes likely increased bacterial presence due to greater surface area.

UV-aged glitter produced ~2× more secondary microplastics compared to pristine samples.

UV and mechanical weathering altered glitter properties, affecting hardness and elastic modulus, with color-dependent responses.

In blue glitter, increasing concentration raised elastic modulus but reduced hardness.

Raman spectroscopy revealed material differences between glitter colors — explaining variation in responses to aging and fragmentation — and chemical differences in the glitters after UV aging.

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