

Sample Preparation

Degra

light dation

Data Collection



IMPROVING DEGRADED MICROPLASTIC IDENTIFICATION USING FTIR-BASED CNN MODELING

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Model Training

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INTRODUCTION:

Identifying plastics in the environment is difficult as their chemical compositions change due to ultra-violet (UV) exposure. This work captures the degradation of plastics in a controlled environment and trains a Convolutional Neural Network (CNN) model to better identify these plastics.

- Fourier Transform Infrared (FTIR) Spectroscopy: Uses infrared light which is absorbed at different frequencies in the atomic bonds of a sample. This absorption is observed in spectra data and helps to identify different bonds.
 - Is more sensitive to the oxidation process polymers undergo during the degradation process.

METHODOLOGY:

A dataset focused on these eight plastics and polyurethane is created with FTIR spectra data from the FLOPP and FLOPP-e library ² and sample scans.

- # of spectrums used: 363
- Spectra range used: 675-4000 cm⁻¹

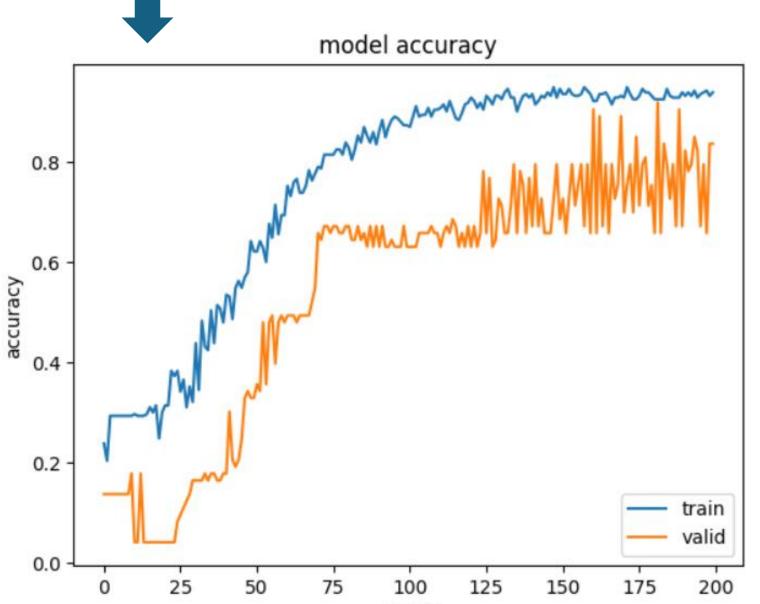
This data is used to train a 1-D CNN model. The CNN structure used is shown above ³.

RESULTS:

Below showcases the split in validation (20%) and training sets.

history=model.fit(x_train, y_train, batch_size=64, epochs=200, validation_split = 0.2, verbose=1)

Below are the training and validation accuracy curves graphed.



Training: How well the model performs on training data.

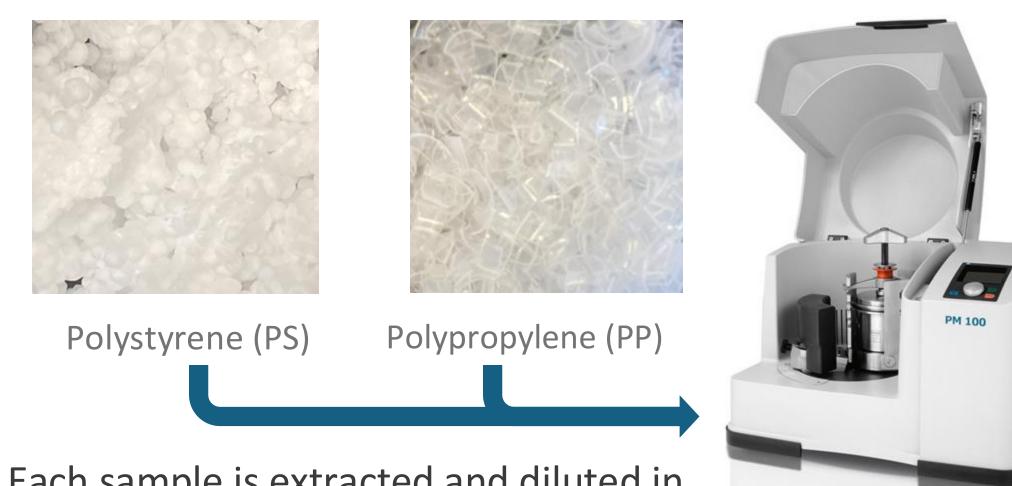
Validation: How well the model performs on new data.

Epochs: How many times the model looks at the data.

Training Accuracy: 94.6% Validation Accuracy: 83.6%

METHODOLOGY

Samples are prepared into 2-5mm sample sizes and then grinded down in a Planetary Ball Mill PM 100¹.



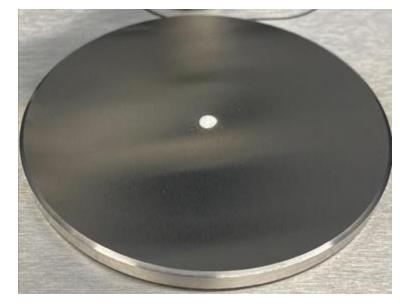


QUV Accelerated Weather Tester

and undergo centrifugation at 3000-3500 rpm. 3-10µL of concentrated



sample is dried and scanned using FTIR Spectroscopy.



Fisher Scientific Accuspin 8C Clinical Centrifuge

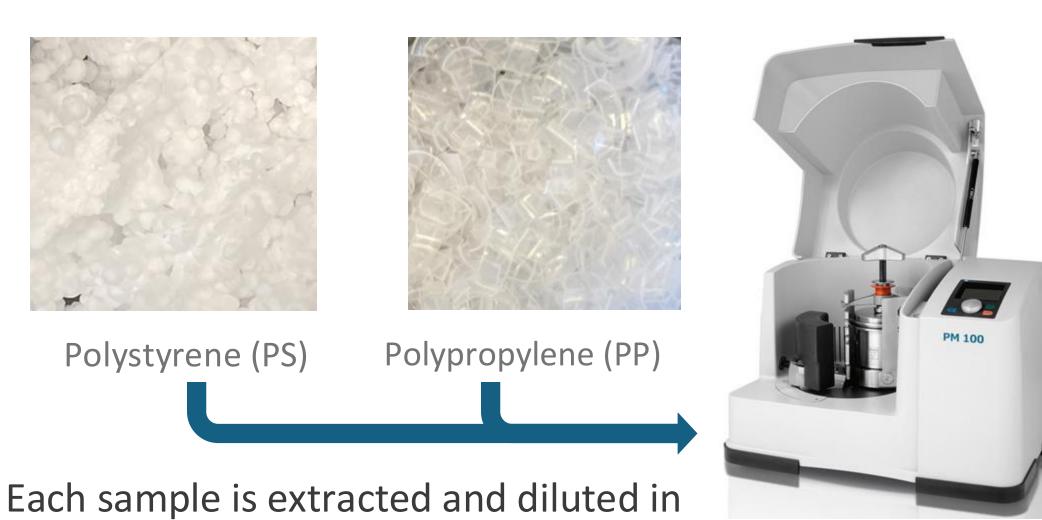
Stage of the FTIR

Additional samples are scanned including:

- polystyrene (PS)
- polyethylene terephthalate (PET) polyethylene (PE)

nylon (PA)

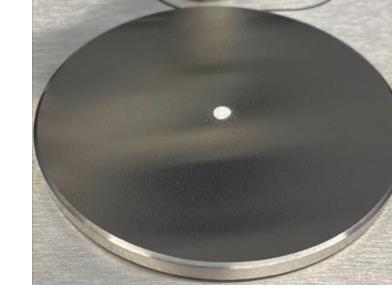
- polypropylene (PP)
- polycarbonate (PC)
- polyvinyl chloride (PVC) polyester



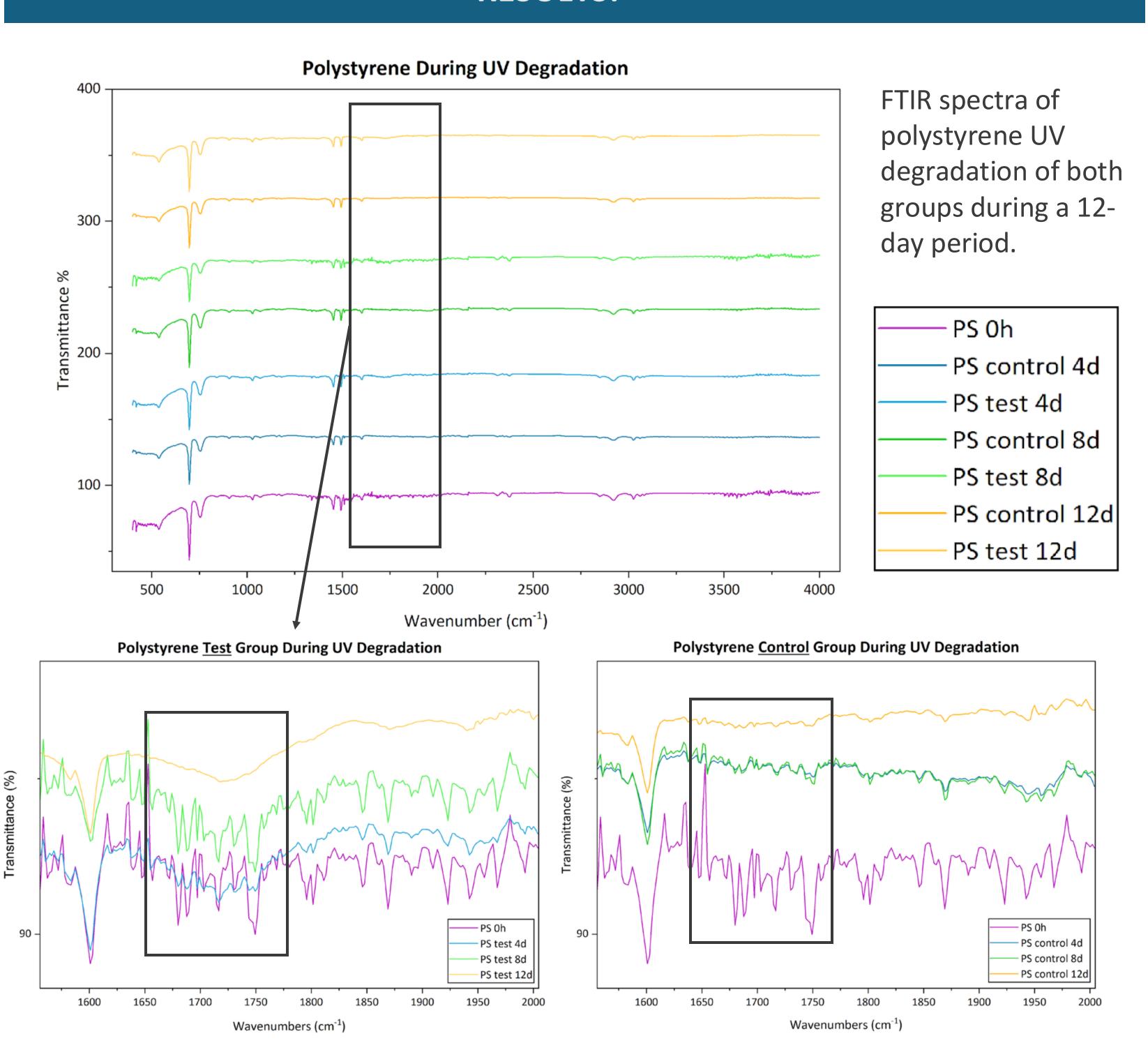
two beakers with 300mL of deionized water.



Samples are degraded in a UV weathering chamber, at an intensity of 340 UVA. Samples are taken out every 4 days



RESULTS:



The peak at ~1725 cm⁻¹ indicates a carbonyl bond is present, a sign of oxidization which is not present in the control group. This agrees with other literature where carbon-oxygen bonds occur during UV exposure 4 and validates the methodology used.

FUTURE WORK:



- Conduct data augmentation to further improve accuracy
- Collect more data on different plastics (polyurethane)
- Further degrade existing samples and different plastics
- Use a saline solution (3.5% sodium) chloride in a water solvent) and study the UV degradation of plastic samples in saltwater environments (left).

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